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A Study of Lewis Acidity in Boron

Kerri McKeever

Supervisors: Prof. Małgorzata (Gosia) Swadźba-Kwaśny, Prof.
John Holbrey, Dr. Sofia Diaz-Moreno, Dr. Daniel Bowron

Confidential

QUILL meeting, 26th March 2024

Abbreviations



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XRS – X-ray Raman Spectroscopy

XAS – X-ray Absorption Spectroscopy

NEXAFS – Near Edge X-ray Absorption Fine Structure

FLP – Frustrated Lewis Pair

FIA – Fluoride ion affinity

HIA – Hydride ion affinity

LUMO – Lowest Unoccupied Molecular Orbital

Presentation Overview



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Background

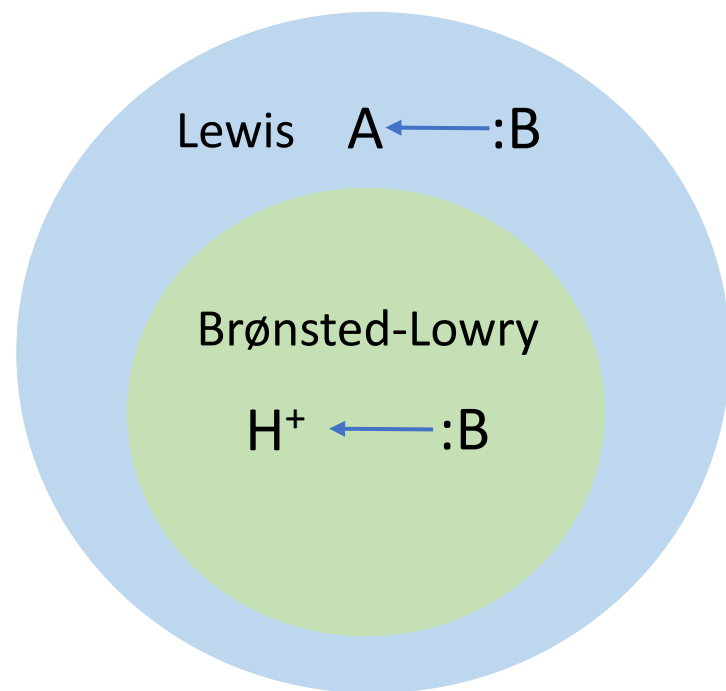
Introduction to XRS

XRS & XAS data

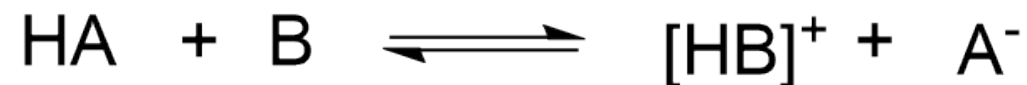
Conclusion and Future Work

Background

- Quantifying Lewis acidity



Brønsted Acidity:



Lewis Acidity:

	NMR	Computational
Hard	 Gutmann-acceptor number method	Fluoride ion affinity (F ⁻)
Soft	 Childs method	Hydride ion affinity (H ⁻)

Background

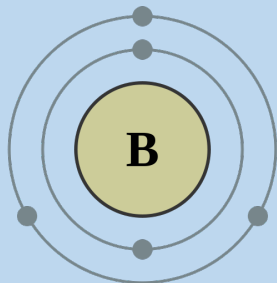
- Factors that affect Lewis acidity:



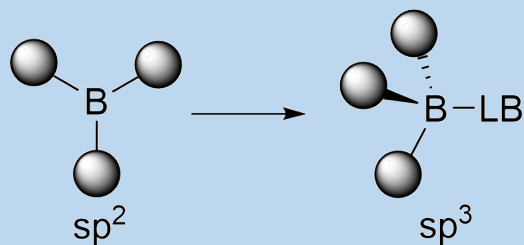
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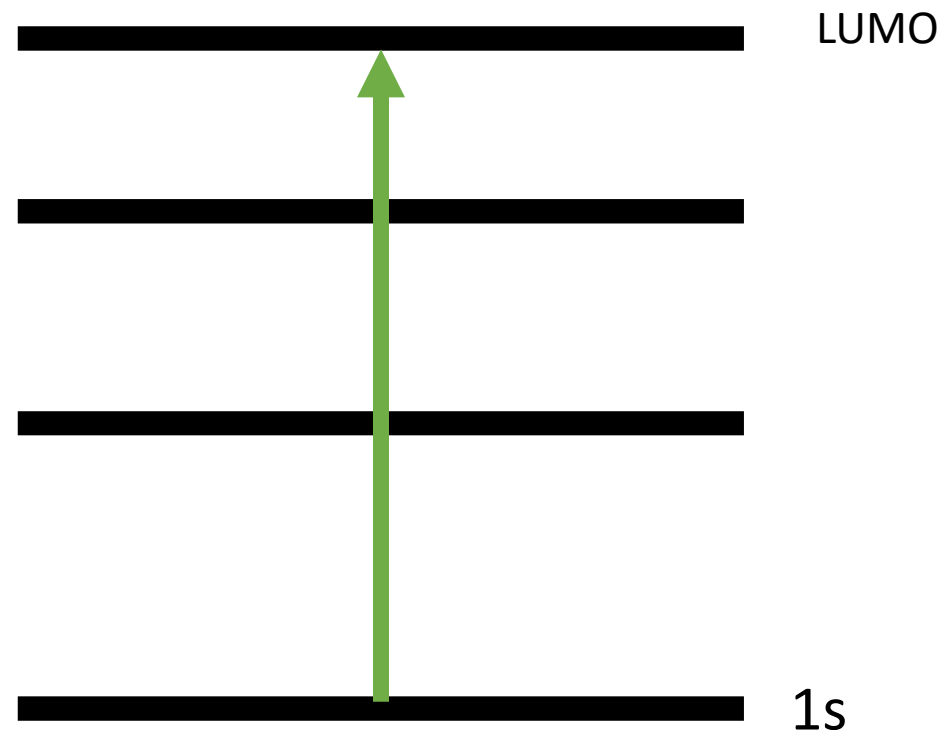
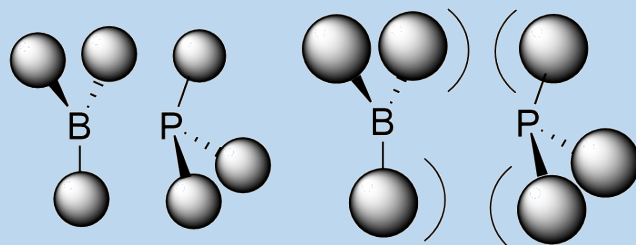
Electrotonic
structure of
boron centre



Energy required
to change
geometry



Steric
hindrance



Background

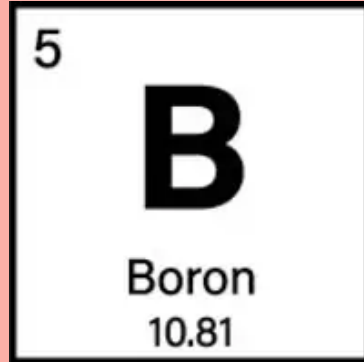
- Challenges with this project:



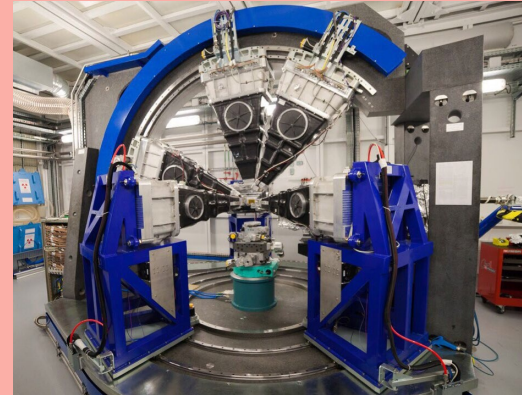
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X-ray absorption
requires ultra-high vacuum



X-ray Raman
niche technique



Background

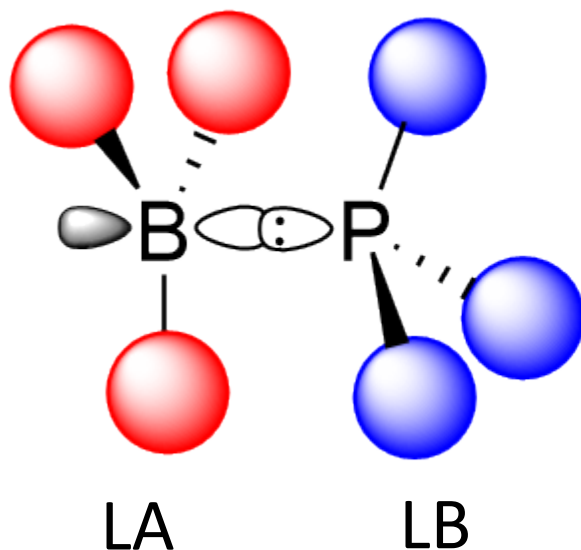
- Frustrated Lewis Pairs



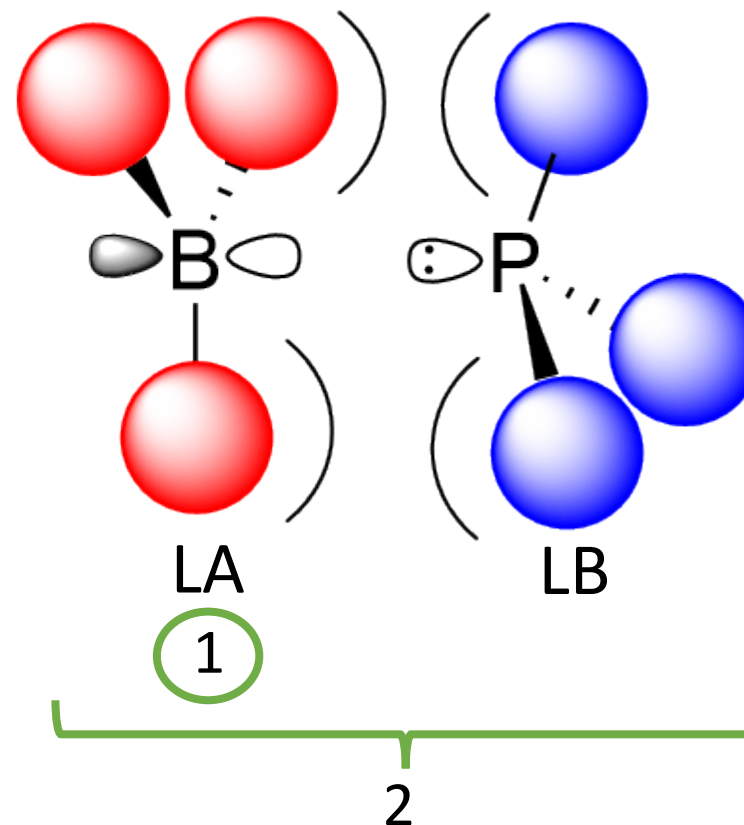
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Classical
adduct



Frustrated
Lewis Pair



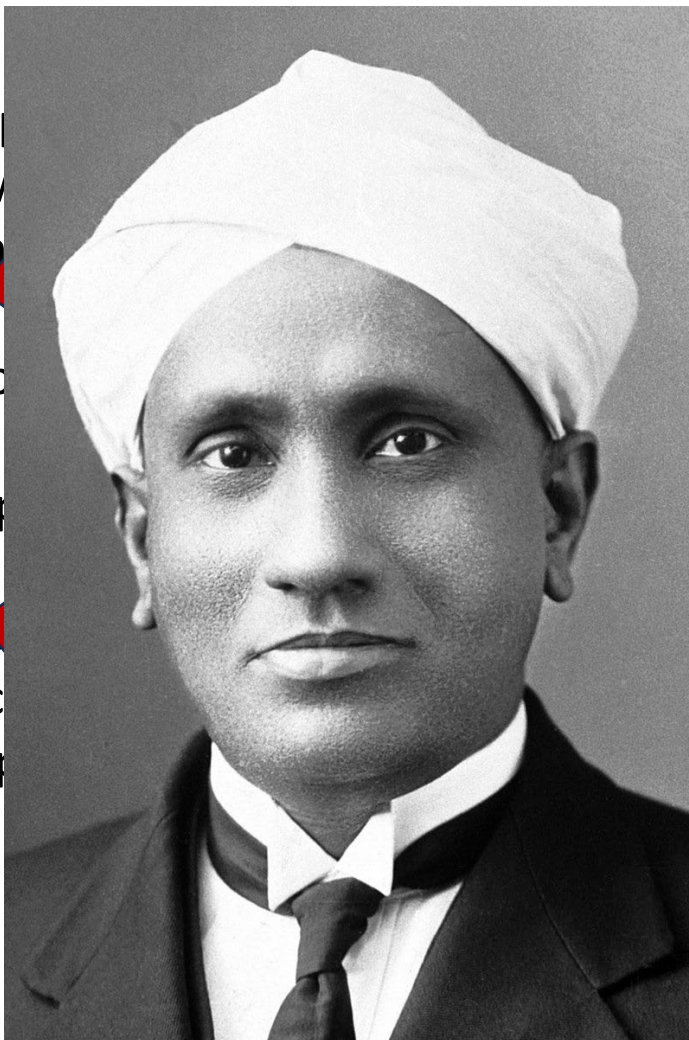
Ramen



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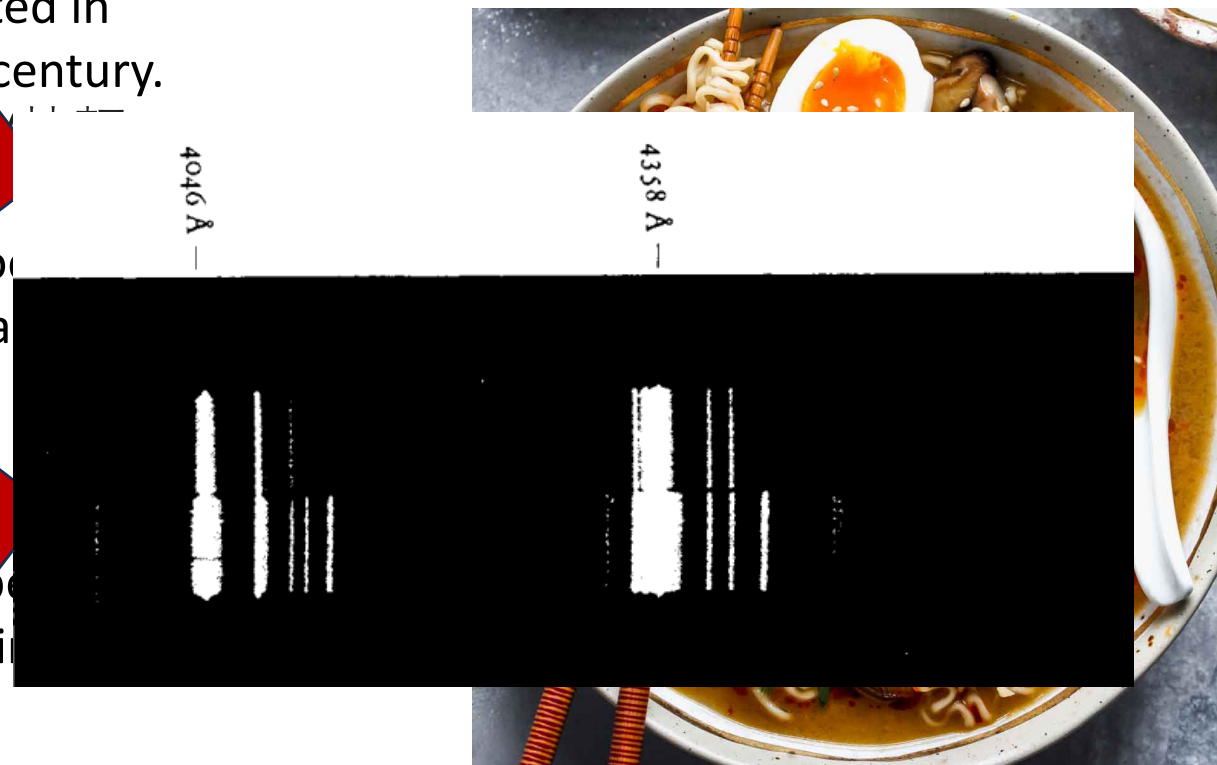
- Ramen is a Japanese dish that originated in China but was popularized in Japan in the early 20th century.
- Ramen comes from the Chinese word 'ramen' meaning 'noodle'.
- The first ramen restaurant in Japan was opened in 1859, when a Chinese immigrant, Wang Gongliang, opened a restaurant in Yokohama, Japan.
- Ramen quickly became a popular dish in Japan, and even more popular during the Second World War for soldiers.



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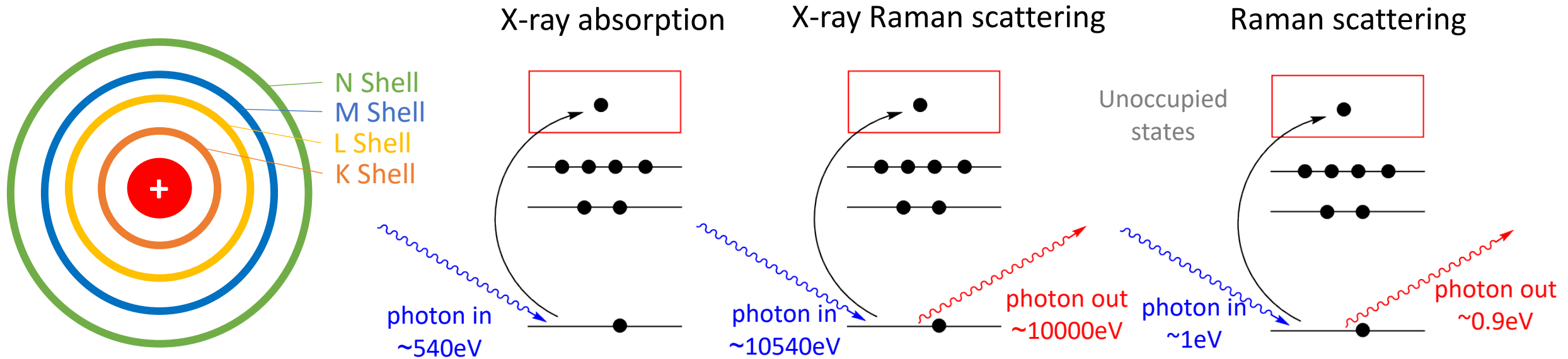


X-ray Raman Spectroscopy (XRS)



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X-ray Raman Spectroscopy (XRS)



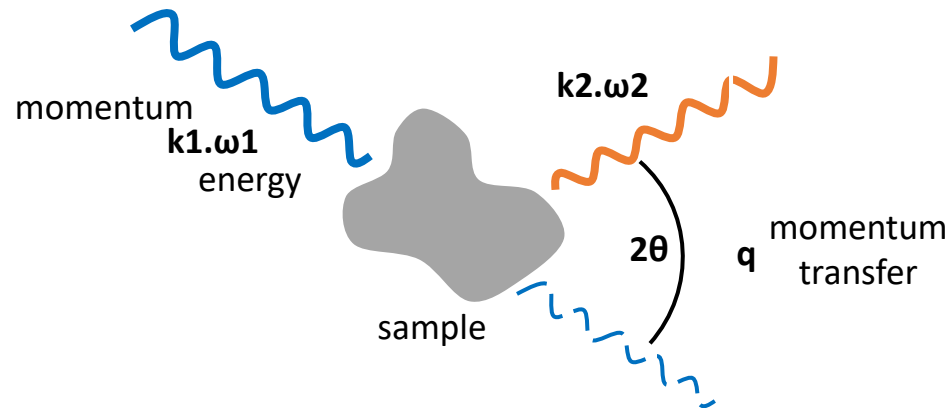
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A few points to consider...

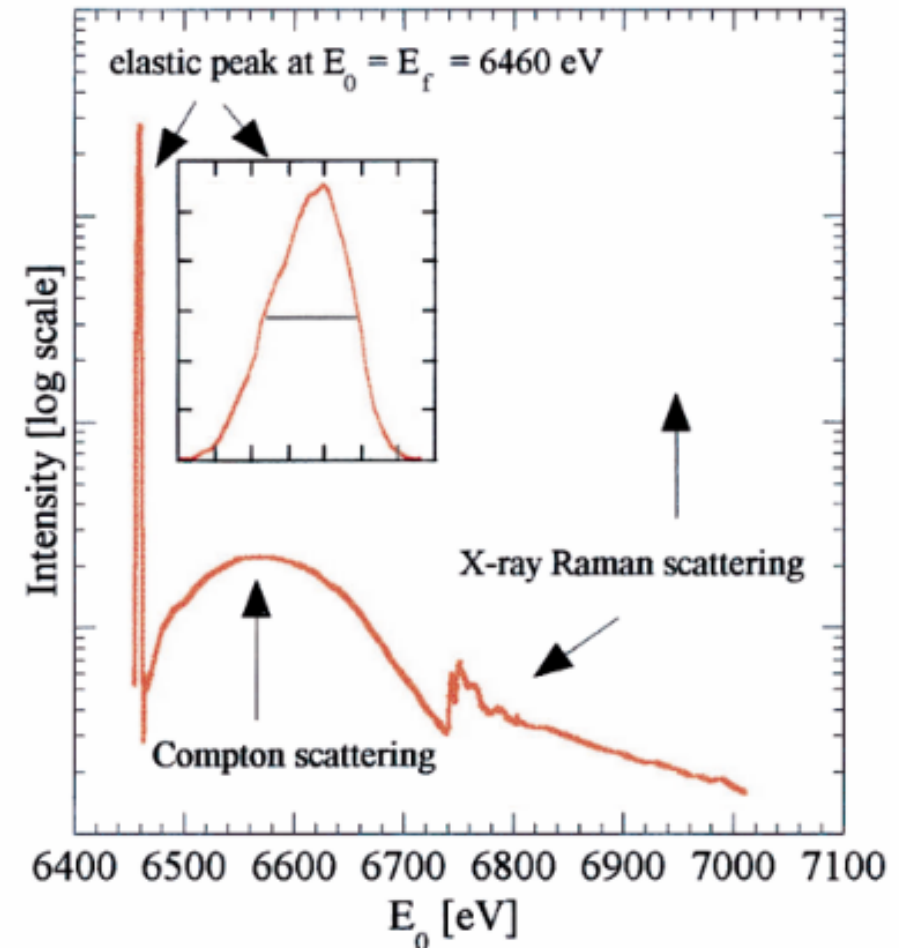
Is it equivalent to XAS?

- Only at low momentum transfer (q) where dipole transitions dominate



Why not use soft XAS?

- Sample environment – vacuum
- Bulk-sensitive vs surface-sensitive

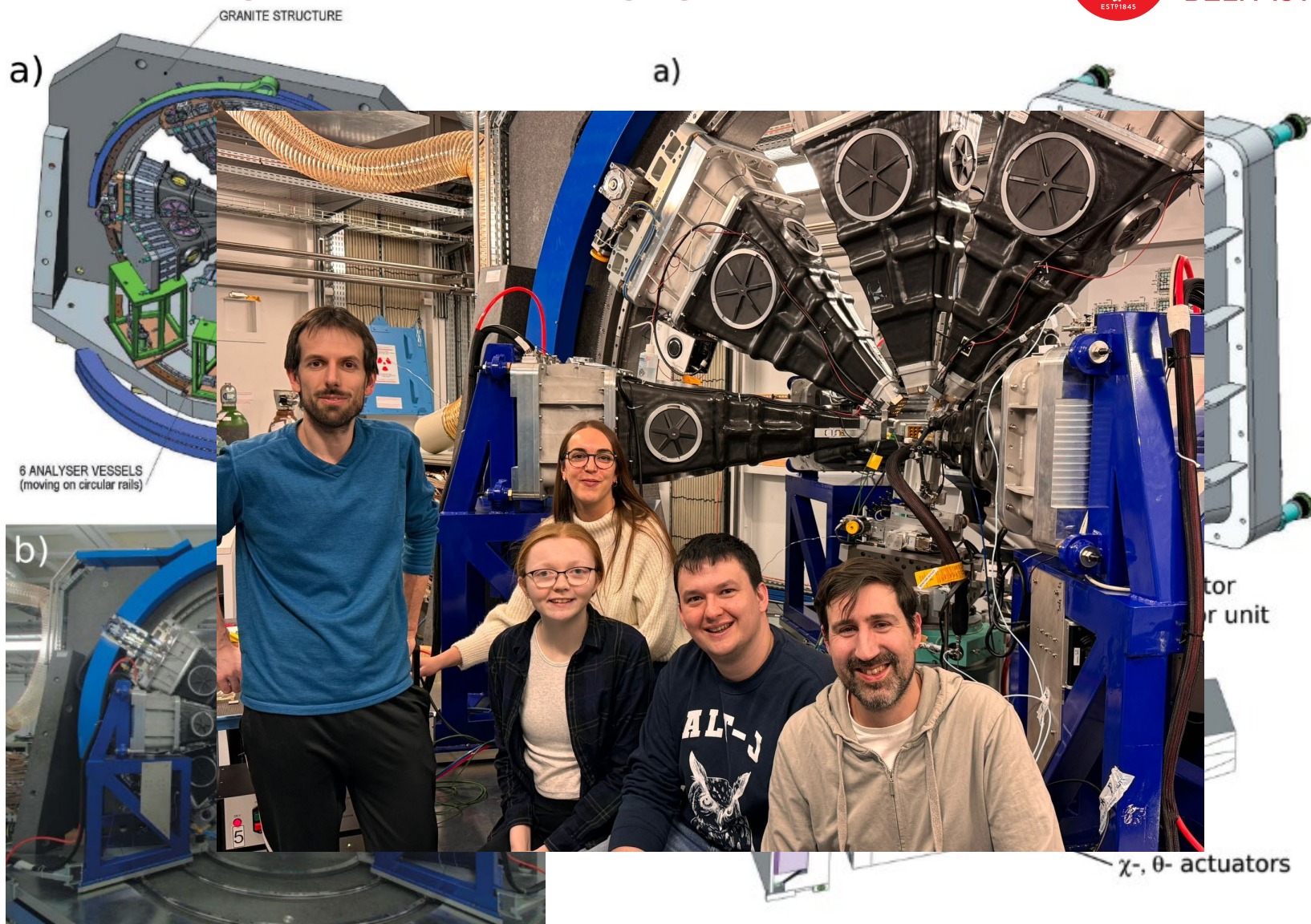


X-ray Raman Spectroscopy (XRS)



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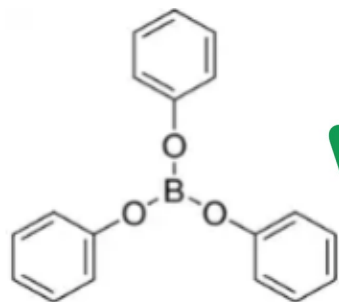


XRS – compounds to analyse

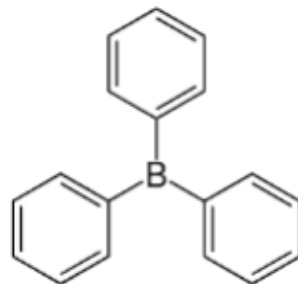


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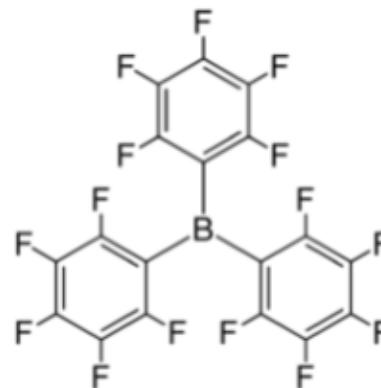
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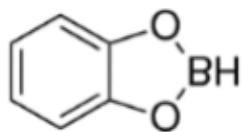
B(OPh)₃ – solid



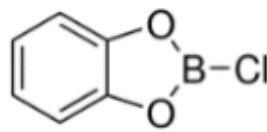
B(Ph)₃ - liquid



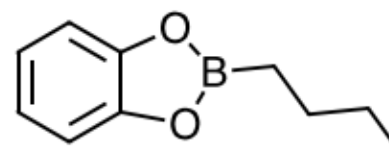
BCF – solid



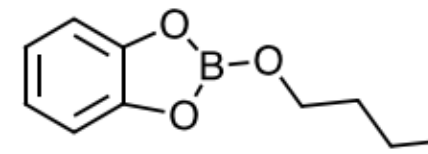
B(cat)H - solid



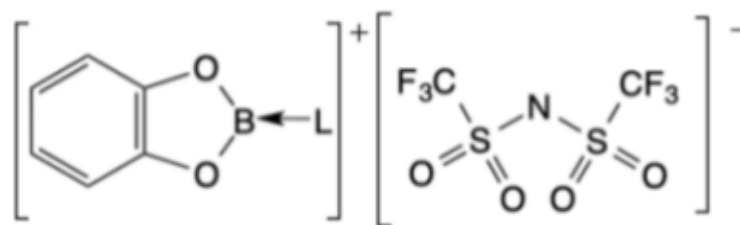
B(cat)Cl - solid



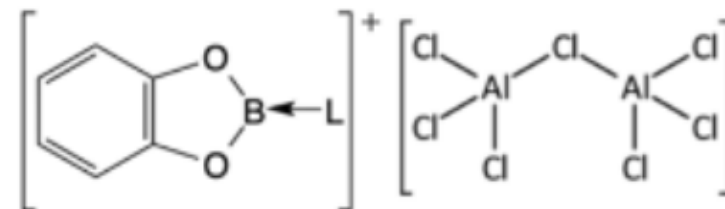
B(cat)Bu - liquid



B(cat)OBu - liquid



[B(cat)L][NTf₂] - liquid



[B(cat)L][Al₂Cl₇] - liquid

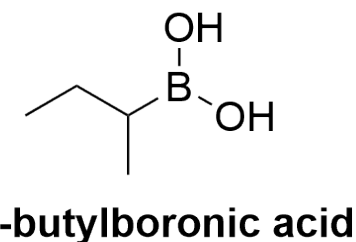
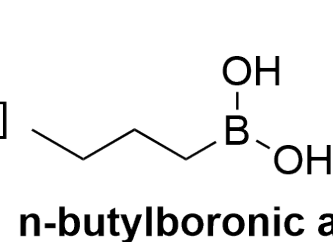
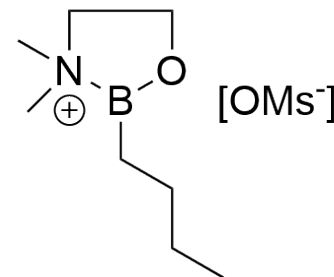
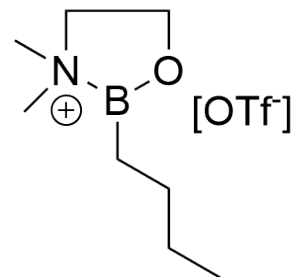
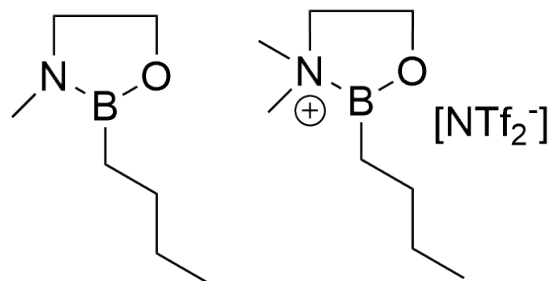
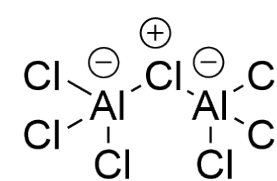
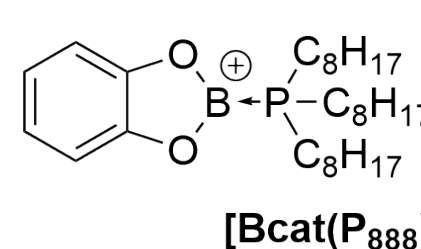
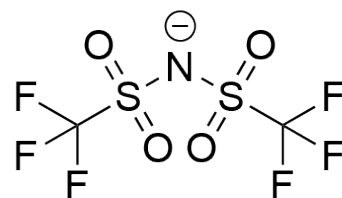
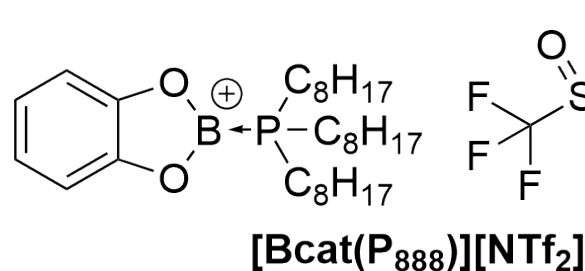
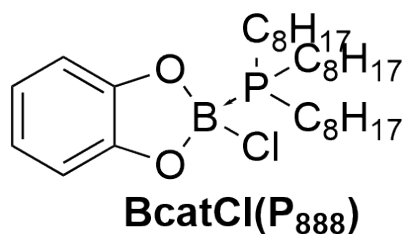
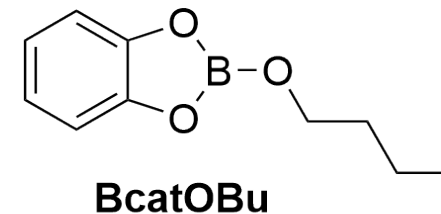
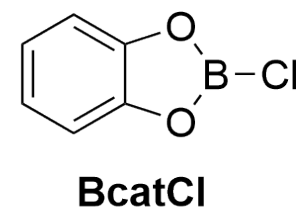
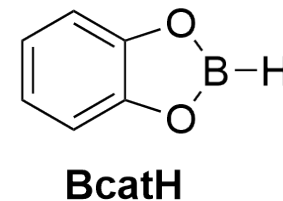
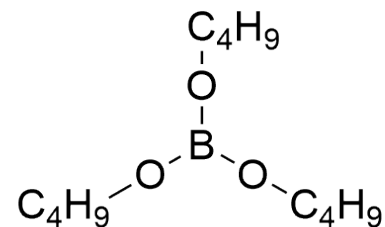
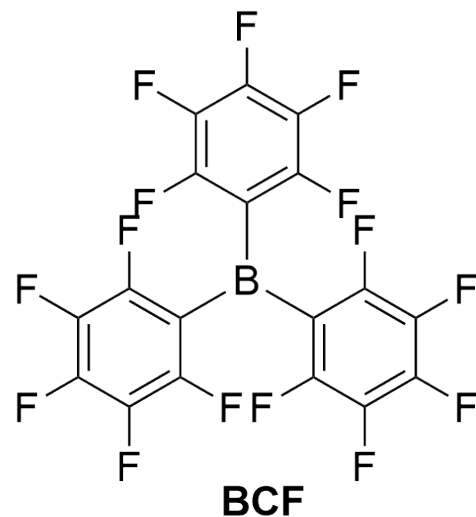
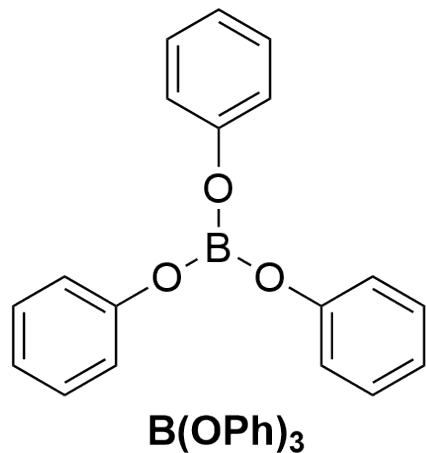
L= 4-picoline

XAS – compounds to analyse



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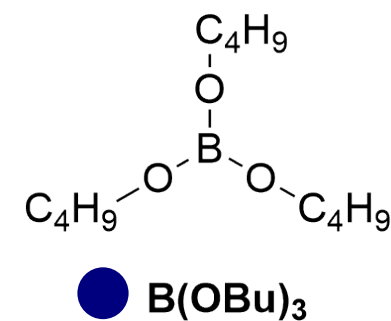
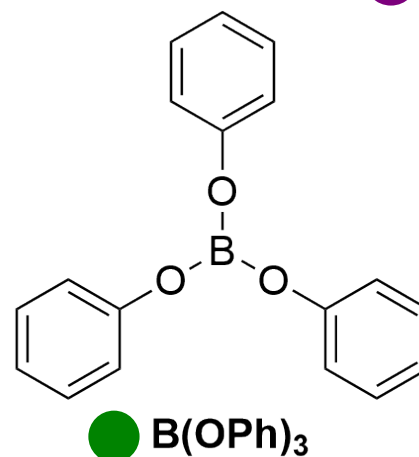
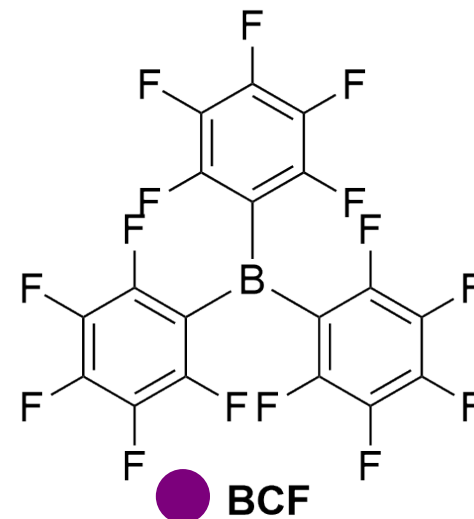
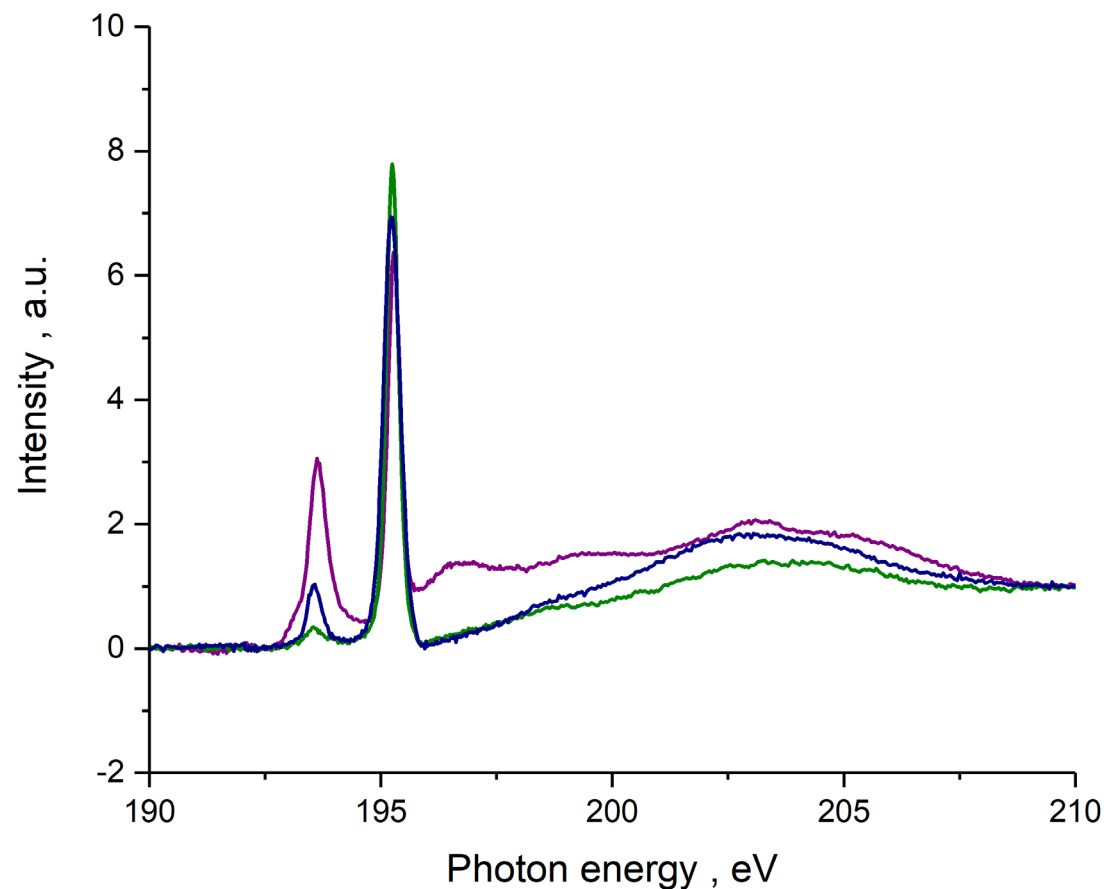


XAS results – B K-edge



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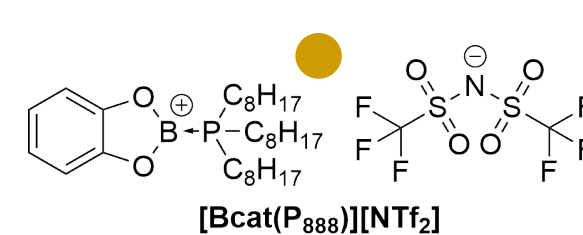
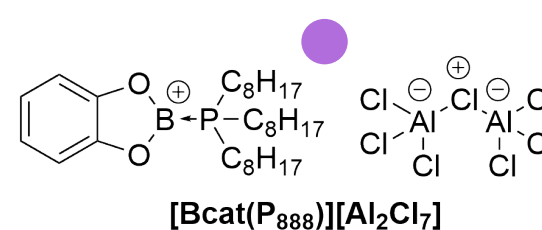
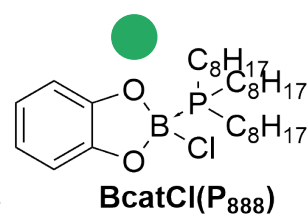
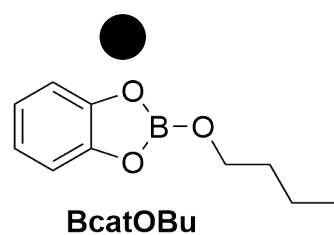
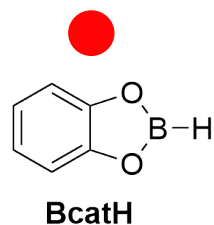
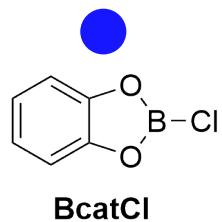
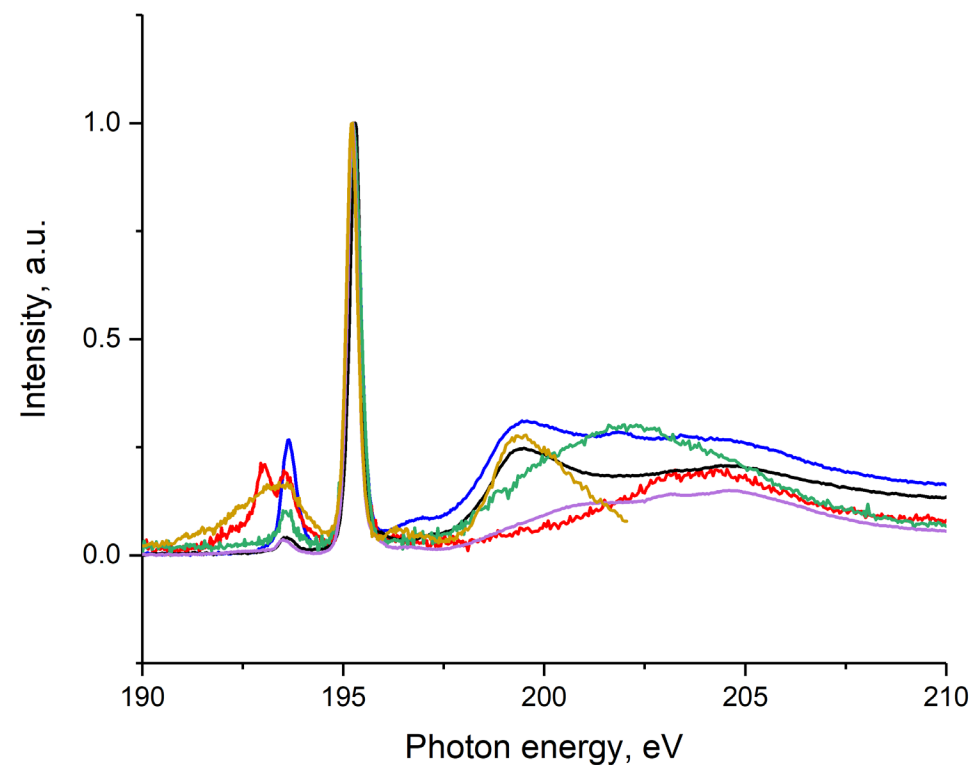
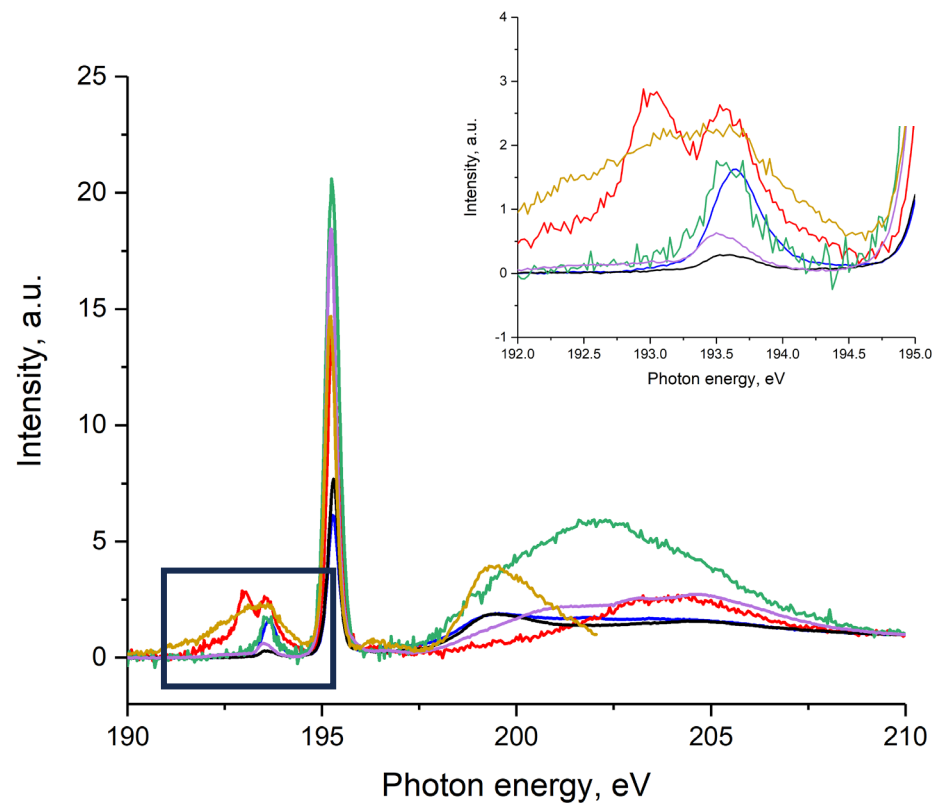


XAS results – B K-edge



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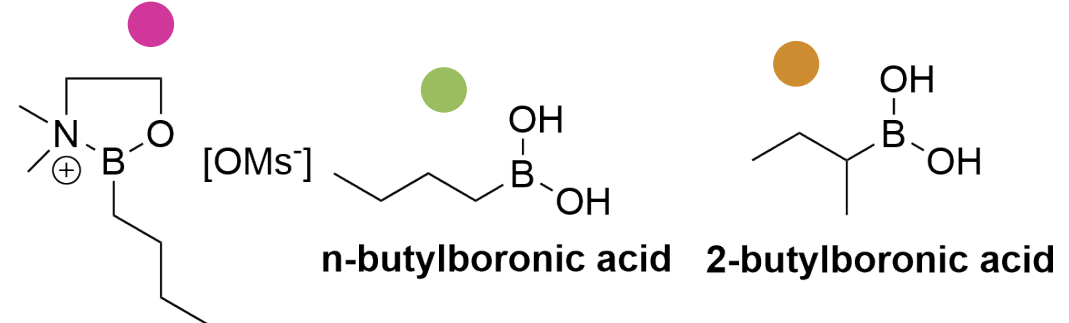
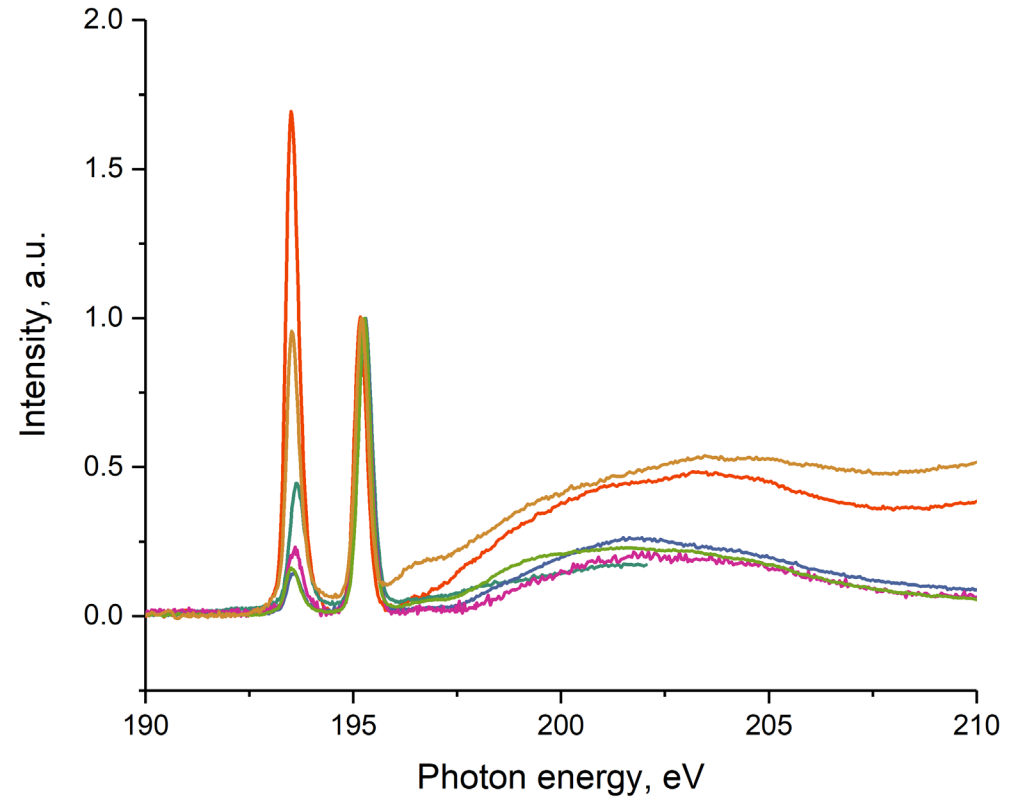
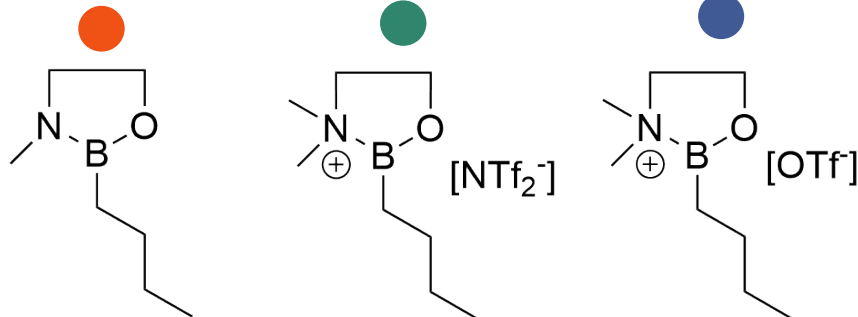
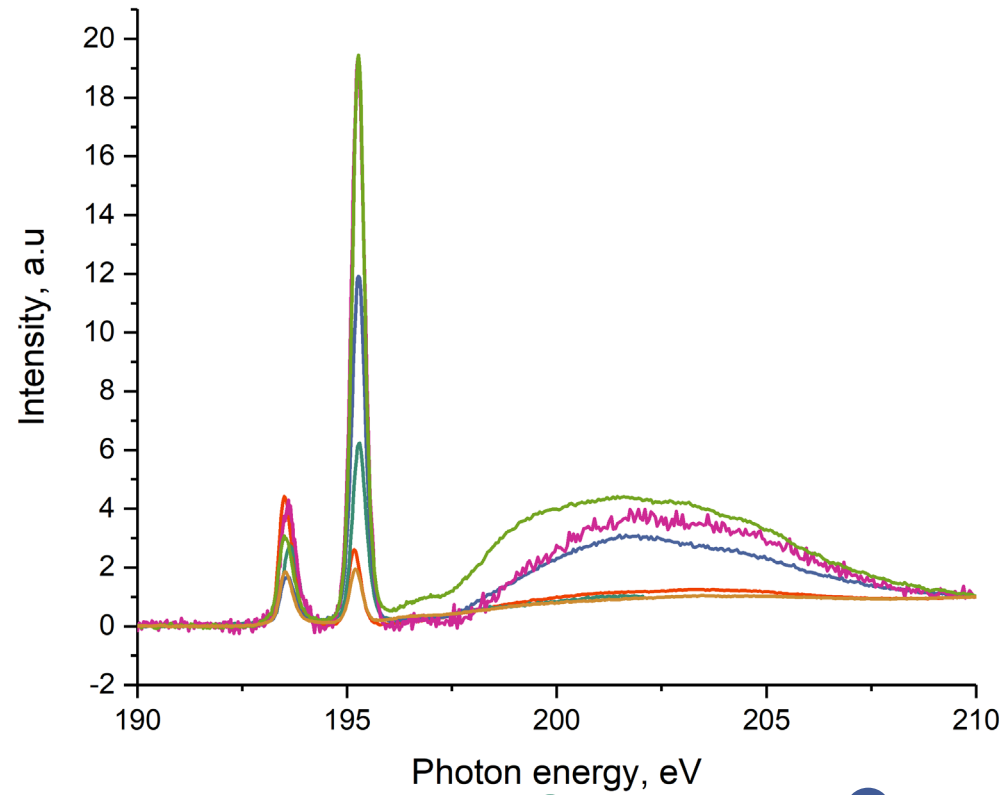


XAS results – B K-edge



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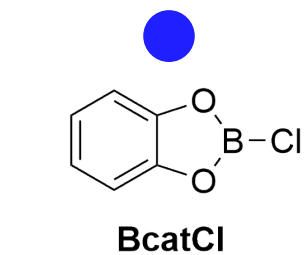
XAS & XRS – B K-edge



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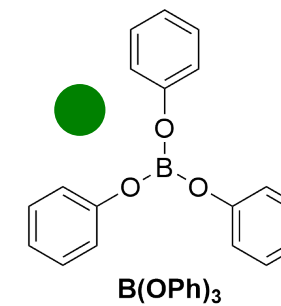
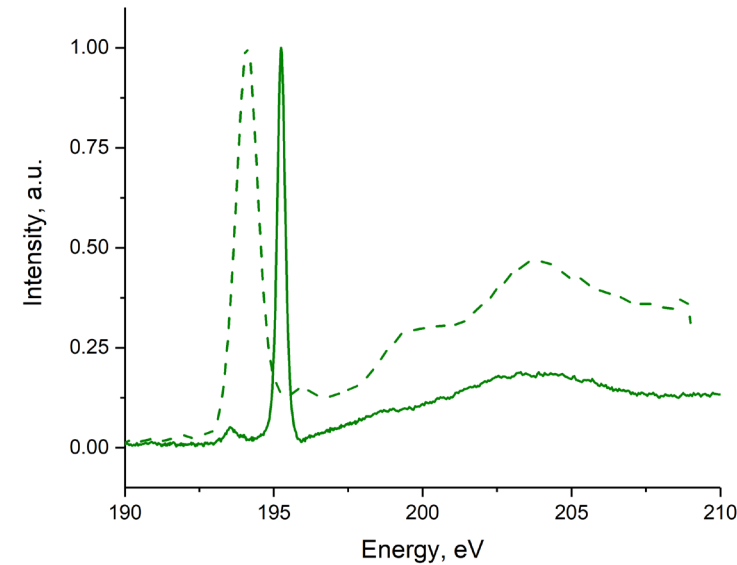
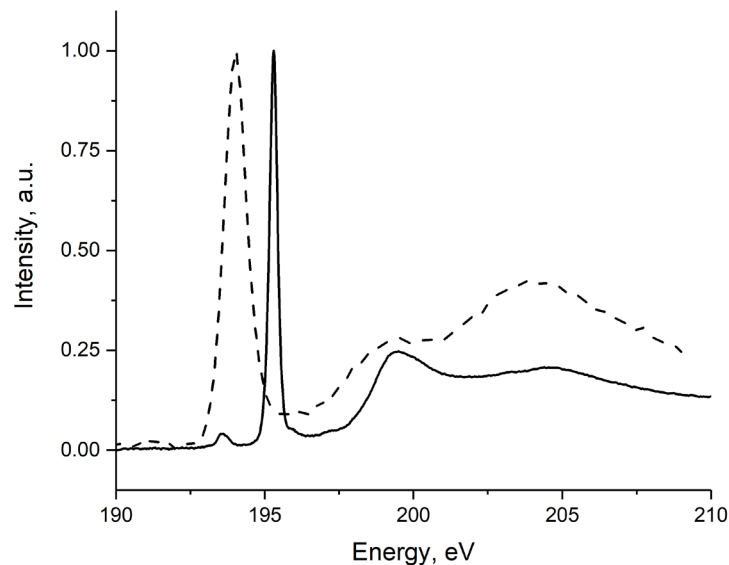
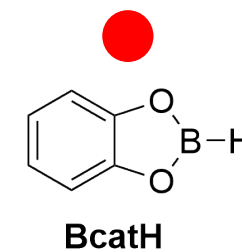
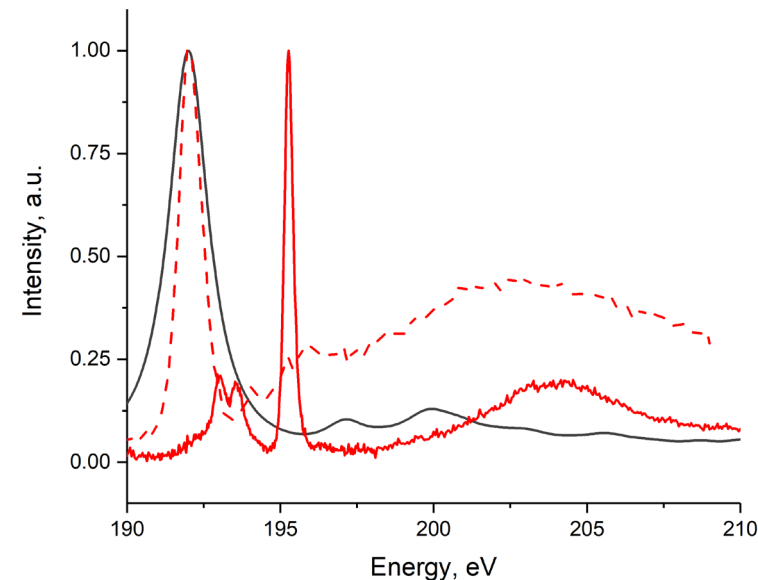
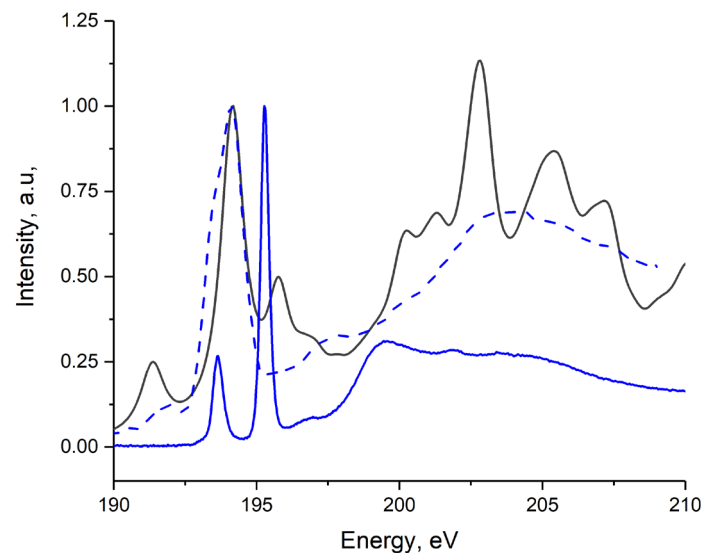
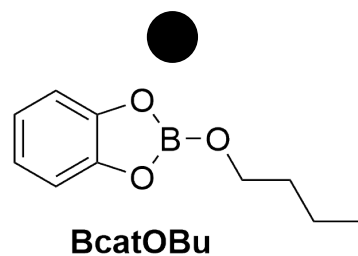
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— XAS
- - - XRS

— simulation



Summary and Future Work



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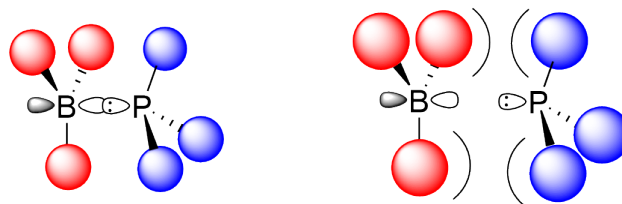
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Summary:

- Measured the B K edge of a range of boron Lewis acids using XAS and XRS to measure the $1s \rightarrow \text{LUMO}$ transition to isolate a component of Lewis acidity

Future work:

- Continue with XRS and XAS analysis
- Applied for beamtime at ESRF to study the B K edge of FLPs
- Conduct operando studies of hydrogenation reactions with FLP-SILP systems.



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- Prof. John Holbrey
- Dr. Sofia Diaz-Moreno (Diamond)
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- Dr. Pilar Ferrer (Diamond)
- Dr Josh Elliott (Diamond)
- Dr. Yoan Delavoux
- Miss Lauren O'Neill
- Miss Emma McCrea
- Miss Beth Murray



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